

Reference Tables for Physical Setting/EARTH SCIENCE

Radioactive Decay Data

Radioactive Isotope	Disintegration		Half-Life (years)
	Parent	Daughter	
Carbon-14	$^{14}_6\text{C}$	$^{14}_7\text{N}$	5.7×10^3
Potassium-40	$^{40}_{19}\text{K}$	$^{40}_{20}\text{Ca}$ $^{40}_{18}\text{Ar}$	1.3×10^9
Uranium-238	$^{238}_{92}\text{U}$	$^{206}_{82}\text{Pb}$	4.5×10^9
Radium-226	$^{226}_{88}\text{Ra}$	$^{206}_{82}\text{Pb}$	1.6×10^3

Equations

Exponential = $\frac{\text{distance between two lengths of equal size}}{\text{length of equal size}}$ **High # - Low #**

Gradient = $\frac{\text{change in field value}}{\text{distance}}$

Rate of change = $\frac{\text{change in value}}{\text{time}}$

Density = $\frac{\text{mass}}{\text{volume}}$

Specific Heats of Common Materials

MATERIAL	SPECIFIC HEAT (Joules/kg \cdot °C)
Liquid water	4.18
Solid water (ice)	2.09
Water vapor	2.00
Dry air	1.01
Steel	0.50
Granite	0.79
Iron	0.45
Copper	0.38
Lead	0.13

Properties of Water

Heat energy gained during melting	334 J/g
Heat energy released during freezing	334 J/g
Heat energy gained during vaporization	2260 J/g
Heat energy released during condensation	2260 J/g
Density at 4.0°C	1.0 g/cm 3

Average Chemical Composition of Earth's Crust, Hydrosphere, and Troposphere

ELEMENT (Symbol)	CRUST		HYDROSPHERE	TROPOSPHERE
	Percent by mass	Percent by volume	Percent by volume	Percent by volume
Oxygen (O)	46.6	44.4		
Silicon (Si)	28.2	27.6		
Aluminum (Al)	8.1	7.8		
Iron (Fe)	5.6	5.4		
Calcium (Ca)	4.1	3.9		
Sodium (Na)	2.8	2.7		
Magnesium (Mg)	2.3	2.2		
Potassium (K)	2.0	1.9		
Nitrogen (N)				78.1
Hydrogen (H)			89.0	
Other	0.9	0.8	1.0	1.9

Most Abundant in Crust

Most Abundant in Troposphere

Most Abundant in Hydrosphere

2015 EDITION

This edition of the Earth Science Reference Tables should be used in the classroom beginning in the 2015-16 school year. The first examination for which these tables will be used is the January 2016 Regents Examination in Physical Setting/Earth Science.



Earth Science Annotated

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BURCKLEY; CLIFFORD R. PHILLIPS;
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(Jr.), 1973 **Report** National Research Council (U.S.). Division of Earth Sciences, 1930 **Earth Science** Gene Thomas
Fox, 1964 **Earth Science Digest**, 1948 United States Code Annotated United States, 2007 Lawyers' Reports
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1. Understanding the eBook Earth Science Annotated
 - The Rise of Digital Reading Earth Science Annotated
 - Advantages of eBooks Over Traditional Books
2. Identifying Earth Science Annotated
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Earth Science Annotated
 - User-Friendly Interface
4. Exploring eBook Recommendations from Earth Science Annotated
 - Personalized Recommendations
 - Earth Science Annotated User Reviews and Ratings
 - Earth Science Annotated and Bestseller Lists
5. Accessing Earth Science Annotated Free and Paid eBooks
 - Earth Science Annotated Public Domain eBooks
 - Earth Science Annotated eBook Subscription Services
 - Earth Science Annotated Budget-Friendly Options

6. Navigating Earth Science Annotated eBook Formats
 - ePub, PDF, MOBI, and More
 - Earth Science Annotated Compatibility with Devices
 - Earth Science Annotated Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Earth Science Annotated
 - Highlighting and Note-Taking Earth Science Annotated
 - Interactive Elements Earth Science Annotated
8. Staying Engaged with Earth Science Annotated
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Earth Science Annotated
9. Balancing eBooks and Physical Books Earth Science Annotated
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Earth Science Annotated
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Earth Science Annotated
 - Setting Reading Goals Earth Science Annotated
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Earth Science Annotated
 - Fact-Checking eBook Content of Earth Science Annotated
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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