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H A J I N

# Bridegroom Stories

**SA Adler**



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web if you start with 14 82 g of ca oh<sub>2</sub> c a o h<sub>2</sub> and 16 35 g of h<sub>2</sub>so<sub>4</sub> h<sub>2</sub> s o<sub>4</sub> a determine the limiting reagent b determine the number of moles of h<sub>2</sub>o h<sub>2</sub> o produced c determine the number of grams of caso<sub>4</sub> c a s o<sub>4</sub> produced d determine the number of grams of excess reagent left 1 make sure the equation is balanced

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