

# Chemistry Study Guide

Periodic Table of the Elements

|    |    |    |    |    |    |    |    |    |    |    |    |     |    |     |    |     |     |   |    |    |    |    |    |    |    |    |    |
|----|----|----|----|----|----|----|----|----|----|----|----|-----|----|-----|----|-----|-----|---|----|----|----|----|----|----|----|----|----|
| 1  | 2  |    |    |    |    |    |    |    |    |    |    | 3   | 4  | 5   | 6  | 7   | 8   | 9 | 10 | 11 | 12 | 13 | 14 | 15 | 16 | 17 | 18 |
| H  | He |    |    |    |    |    |    |    |    |    |    | B   | C  | N   | O  | F   | Ne  |   |    |    |    |    |    |    |    |    |    |
| Li | Be |    |    |    |    |    |    |    |    |    |    | Al  | Si | P   | S  | Cl  | Ar  |   |    |    |    |    |    |    |    |    |    |
| Na | Mg | Sc | Ti | V  | Cr | Mn | Fe | Co | Ni | Cu | Zn | Ga  | Ge | As  | Se | Br  | Kr  |   |    |    |    |    |    |    |    |    |    |
| K  | Ca | Y  | Zr | Nb | Mo | Tc | Ru | Rh | Pd | Ag | Cd | In  | Sn | Sb  | Te | I   | Xe  |   |    |    |    |    |    |    |    |    |    |
| Rb | Sr |    | Hf | Ta | W  | Re | Os | Ir | Pt | Au | Hg | Tl  | Pb | Bi  | Po | At  | Rn  |   |    |    |    |    |    |    |    |    |    |
| Cs | Ba |    | Rf | Db | Sg | Bh | Hs | Mt | Ds | Rg | Cn | Uut | Fl | Uup | Lv | Uus | Uuo |   |    |    |    |    |    |    |    |    |    |
| Fr | Ra |    | Rf | Db | Sg | Bh | Hs | Mt | Ds | Rg | Cn | Uut | Fl | Uup | Lv | Uus | Uuo |   |    |    |    |    |    |    |    |    |    |
|    |    | La | Ce | Pr | Nd | Pm | Sm | Eu | Gd | Tb | Dy | Ho  | Er | Tm  | Yb | Lu  |     |   |    |    |    |    |    |    |    |    |    |
|    |    | Ac | Th | Pa | U  | Np | Pu | Am | Cm | Bk | Cf | Es  | Fm | Md  | No | Lr  |     |   |    |    |    |    |    |    |    |    |    |

## Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🎯

## Matter:

Anything that has **mass** and **occupies space** 🌟🌍

Examples: air, gases, liquids, solidz!

## Atoms:

Atoms are the **smallest, indivisible particle** 🔬👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons ( $e^-$ )
- **Inner part** → *nucleus* → houses the protons ( $p^+$ ) and neutrons ( $n$ )
- Outer & Inner part → together these are the **subatomic particles**

## Subatomic Particle Table:

| Particle | Symbol | Mass (amu)       | Charge | Location    |
|----------|--------|------------------|--------|-------------|
| Proton   | $p^+$  | 1                | +1     | Nucleus     |
| Neutron  | $n$    | 1                | 0      | Nucleus     |
| Electron | $e^-$  | $\frac{1}{1836}$ | -1     | Orbit/Shell |

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