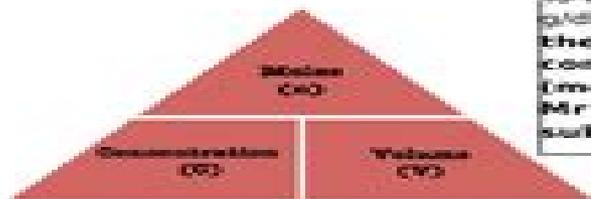


# Chemical calculations



**Units**  
 Mass = g  
 moles = mol  
 Relative formula mass = g mol<sup>-1</sup>

## Moles and concentration

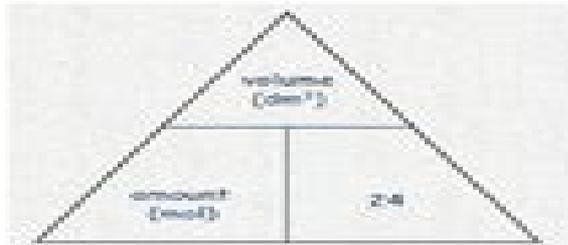


To convert into g/dm<sup>3</sup> multiply the concentration (mol/dm<sup>3</sup>) by the Mr of the substance

$Moles = \frac{Mass}{Relative\ formula\ mass}$   
 $Mass = Moles \times Relative\ formula\ mass$   
 $Relative\ formula\ mass = \frac{Mass}{Moles}$

$1\ dm^3 = 1000\ cm^3$

## Moles and gas volumes



How much we actually get from the reaction

**Percent Yield =  $\frac{Actual\ Yield}{Theoretical\ Yield} \times 100\%$**

How much we predict we will get

## Calculation of Atom Economy

**atom economy =  $\frac{mass\ of\ atoms\ in\ desired\ product}{mass\ of\ atoms\ in\ reactants} \times 100\%$**

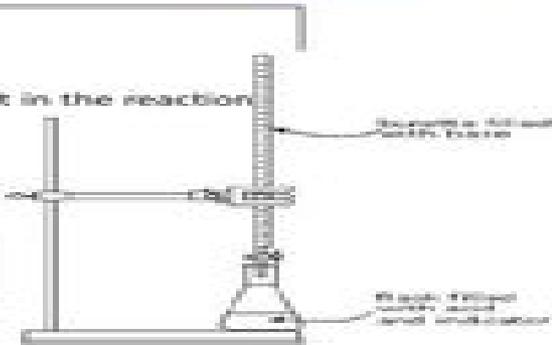
## Titration calculations

- Step 1** - Write out the balanced symbol equation in the question
- Step 2** - Write all the information you have about each reactant/product in the reaction
- Step 3** - Make sure volumes are converted into **dm<sup>3</sup>**
- Step 4** - Calculate number of moles of the known concentration, then you will know moles of unknown.

You can then carry out concentration

Calculation of the unknown.

	HCl	NaOH	NaCl	H <sub>2</sub> O
Amount				
Conc				
Volume				
Concn				
Volume				
Concn				
Mass				
Concn				



# Chemical Calculations

**Jessica J Manson**



## **Chemical Calculations:**

Chemical Calculations Paul Yates, 1997-06-05 *Chemical Calculations* provides an introduction to the mathematics required for physical chemistry courses. This book is unique in that it provides a gentle introduction with a chemistry centered rather than math centered approach. Written by a chemist for undergraduate students it imparts an understanding of the subject from a chemist's viewpoint using examples from real chemistry. It includes illustrations that show exactly how to use calculators to work problems and examples of important chemical problems with fully worked solutions. This book is an ideal companion throughout a chemistry course that can be consulted when required and used to keep one step ahead of the lecture.

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*Chemical Calculations* Paul C. Yates, 2023-04-21 Uniquely organized by chemical rather than mathematical topics this book relates each mathematical technique to the chemical concepts where it applies. The new edition features additional revised and updated material in every chapter and maintains the clarity of the previous edition with the appropriate organization of topics and improved cross referencing where mathematical techniques occur more than once. The text contains additional worked examples and end of chapter exercises with detailed solutions giving students the opportunity to apply previously introduced techniques to chemically related problems. It is an ideal course companion for chemistry courses throughout the length of a degree. Features This book covers the difficult area of mathematics in an easy to read format for students and professionals in chemistry and related subjects. Structured according to chemical rather than mathematical topics. Each topic has at least 12 end of chapter applied chemistry problems to provide practice in applying the techniques to real chemistry. Indexing of material by both chemical and mathematical topics. Extends its utility as a concise and practical reference for professionals in a wide array of scientific disciplines involving chemistry.

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Excerpt from Chemical Calculations With Explanatory Notes Problems and Answers In the division of chemical teaching into departments which sooner or later becomes necessary in a growing College the instruction of students in Chemical Calculations has for several years fallen to the lot of the author of this little treatise Mr Whiteley has also at the same time been engaged in laboratory teaching and in lecturing upon certain special branches of Chemistry The text book which he now issues accordingly bears manifest evidence not only of his experience as a teacher but also of his being well in touch with the calculations which are required by chemical students for various purposes and with the chemical knowledge which elementary students may be assumed to possess As a matter of fact however the author has wisely decided to give a short summary of chemical facts or processes before proceeding to explain the methods of calculation which are applied to them Laboratory students will find a most useful and convenient explanation of the methods of calculating the results of specific gravity determinations of analyses whether gravimetric volumetric or gasometric and of atomic and molecular weight determinations they will find these also brought up to date as regards the methods to which they refer About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books Find more at [www.forgottenbooks.com](http://www.forgottenbooks.com) This book is a reproduction of an important historical work Forgotten Books uses state of the art technology to digitally reconstruct the work preserving the original format whilst repairing imperfections present in the aged copy In rare cases an imperfection in the original such as a blemish or missing page may be replicated in our edition We do however repair the vast majority of imperfections successfully any imperfections that remain are intentionally left to preserve the state of such historical works

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branch

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account whether saving or current in the branch mention the branch number and its location

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papers dear contact name this letter is to acknowledge that we our board of directors have has recently approved the  
addition of a new signing officer to our account no account no

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987766 on my behalf this authorization is effective from start date to end date or no date sincerely

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