

Bioengineering Thermodynamics

Course Objective: Understand and apply the laws of thermodynamics to biological systems, comprehend principles of chemical and physical equilibria, chemical and enzyme kinematics and their application.

First law of thermodynamics

(Contributed by Sumeet)

Properties of pure substances, equations of state, Mollier diagram, closed system, open system, reversible processes, internal energy and enthalpy steady-flow engineering devices and transient flow analysis.

Second Law of Thermodynamics

(Contributed by Sumeet)

Statements of the second law, heat engines, reversible versus irreversible processes, the Carnot cycle, refrigeration devices, entropy and entropy change, third law, exergy analysis.

Free Energy and Chemical Equilibria

(Contributed by Dr Amjad Ali & Dr. Banibrata Maity)

Gibbs free energy, Helmholtz free energy, physical significance of free energy, Gibbs-Helmholtz equation, application of free energy to gases, concept of spontaneity, partial molar Gibbs energy, chemical potential of multicomponent system, reactions of ideal gases, non-ideal systems, equilibrium and standard Gibbs free energy and Bio-chemical applications of thermodynamics. Law of mass action, The Le-Chatelier principle, Van't Hoff reaction isotherm and equations.

Free Energy and Physical Equilibria

(Contributed by Dr Amjad Ali & Dr. Banibrata Maity)

Concepts and applications, phase equilibria for single and multi-component systems, membranes colligative properties and application of thermodynamics to phase transition

Chemical Kinetics

(Contributed by Dr Amjad Ali, Dr. Banibrata Maity & Dr Sudhir Singh)

Concepts and applications, rate of reaction, measurement of rate of reaction, factors influencing rate of reaction, rate laws order and molecularity, integrated rate equations and half-lives, zero order reactions, first order reactions, second order reaction, third order reaction, higher nth order reactions, pseudo-order reactions, temperature dependence of rate of reactions, temperature coefficient, activation energy, Arrhenius equation, rate calculation, collision theory transition state theory.

Enzyme Kinetics

(Contributed by Dr Diptiman)

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AW Rasmussen



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